Complexity of mixed allotropes of MoS₂ unraveled by first-principles theory

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Two-dimensional MoS₂ forms stable and several metastable allotropes of semimetallic, metallic, and semiconducting characters, depending on the experimental growth conditions. In this paper we consider intergrowth of the two most frequently reported metastable phases of MoS_2 (1T and 1T') within the stable 1H phase to establish the effect of geometric and electronic reconstructions of the interface region between 1H and T/1T'phases using first-principles density functional calculations. We show that a complex structural reconstruction at the interfaces is responsible for the opening of an energy gap driving the electronic and geometric structures to resemble that of the 1T' phase. It is also found that the size of the patches inside the 1H matrix crucially controls the geometry and electronic structure close to the Fermi level. These results establish that remarkable properties of chemically exfoliated MoS₂ with patches of metastable structures are to be understood as arising fundamentally from intergrowths necessarily strained due to lattice mismatch across the interface between the H and the distorted T phase, rather than in terms of any pure metastable phase.

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I. INTRODUCTION

The experimental realization of graphene in 2004 [1] boosted the interest on two-dimensional (2D) materials, which can be synthesized by mechanical or chemical exfoliation from van der Waals bonded 2D layers in a bulk phase. Since graphene is a semimetal and thus has a very poor on-off ratio, its range of applicability as an alternative to silicon in semiconductor devices is very limited. One of the options extensively researched in this context is based on 2D transition metal dichalcogenides (TMDs) [2–5].

MoS₂ is the lightest member of the TMD family and it has been studied for decades because it is very abundant in the Earth's crust. It is broadly used as a catalyst [6,7] and solid state lubricant [8,9]. Much like graphite, it is a layered material in which the layers are stacked via van der Waals forces. However, each layer in MoS₂ is in its turn formed by three atomic layers with a thickness of 6.2 Å [10], in which a central molybdenum layer is sandwiched between two sulfur layers. Monolayer MoS₂ is easy to obtain experimentally via mechanical exfoliation. Another route is chemical exfoliation: ion intercalation is possible [11] due to the weak van der Waals forces between the layers, which increases the separation between the layers progressively until the interlayer

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coupling is insignificant [12]. A recent review article [13] discusses the aspects of chemically exfoliated 2D MoS₂.

Bulk MoS₂ has an indirect gap of 1.2 eV [14] but the monolayer is, in its ground state phase, a direct gap semiconductor (1.8 eV) at the high symmetry point K in the Brillouin zone [4]. This, together with its abundance and convenient mechanical properties, has made it interesting for material scientists, leading to exponential increase in the recent publication [11]. Among other applications, 2D MoS₂ is a promising candidate for transparent and flexible electronics [3,15,16].

Monolayer MoS₂ can exist in different structural phases, three of the most commonly discussed in the literature are shown in Fig. 1. The ground state phase is 1H, in which the S and Mo atoms are stacked in an A-B-A fashion, so that S atoms are on top of each other and six S atoms are bonded to the central Mo atom in a trigonal coordination. This phase is, as mentioned before, a direct semiconductor of 1.8 eV band gap, and it has been extensively studied theoretically and experimentally [10,11]. Higher energy polymorphs are rather common as patches in the 1H phase when extracting monolayer MoS₂ via chemical exfoliation, and have an A-B-C stacking with octahedral coordination of S atoms around the central Mo. The pure 1T phase, which can be attained from the 1H via a gliding displacement of one of the S atomic planes, is dynamically unstable. This means that it will spontaneously undergo Jahn-Teller distortions towards one of the stable or metastable phases that have been found so far. The lowest energy metastable phase is called 1T' phase. It is obtained from 1T via the dimerization of the Mo atoms (see Fig. 1), so that there is an alternation of short and long Mo-Mo bonds, which lowers the energy. While the above discussion relates to the pure form of each phase, in practice, the metastable phases invariably appear as small patches intergrown within the stable 1H phase. Consequently, in contrast

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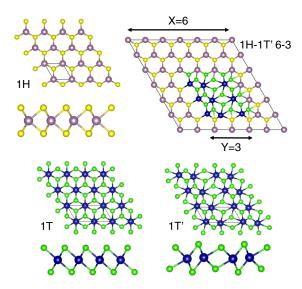


FIG. 1. Three different phases of monolayer MoS₂: the ground state trigonal 1H and the two octahedral polytopes, 1T and 1T'. Mo atoms are shown in purple in 1H and blue in 1T and 1T', while S atoms are yellow in 1H and green in 1T and 1T'. Different colors have been used to distinguish the mixed phase structures in the rest of the paper. An example of these is included in the right top panel: the 1H matrix is a 6×6 supercell, making X = 6, and the octahedral patch is 3×3 , so that Y = 3. Since the patch is in this case 1T', this structure is named 1H-1T' 6-3.

to the considerable research carried out for the 1H phase, fewer studies are available on these octahedral forms, and most of them are theoretical works due to the difficulty to spatially resolve the small patches experimentally with photoemission spectroscopy [17].

The 1T phase is known to be metallic [18], which could be beneficial for some applications such as supercapacitors [19], if there was a way of stabilizing it. In fact, it has been implicitly hoped for that the instability of the pure 1T phase may be alleviated when it is formed as small patches of intergrowth areas within the stable 1H phase. The calculated electronic structure of dimerized 1T' phase exhibits a small gap in the presence of spin-orbit coupling, though the magnitude of this gap somewhat varies across different calculations [2,20]. In a recent paper [2], the metastable trigonal 1T' phase has been observed in experiments within the matrix of hexagonal 1H phase yielding a band gap around 90 meV, supported by theoretical calculations of bulk phase using GW method. It is extremely challenging to experimentally assess which of the metastable phases is found in experiments, because they usually coexist with the 1H phase as small patches, which imply finite size effects, strain, charge doping, etc. This complicates the interpretation of the results and leads to conflicting reported results [13]. Therefore, understanding the structural and electronic properties of these composites is nontrivial. Even theoretically, very little attention has been drawn to the interfaces between these phases while mostly pure phases have been investigated. There are extensive experimental reports on the topic of phase boundaries and phase transitions between the different phases in 2D transition metal dichalcogenides [21-23], and several theoretical works that

describe phase transitions [24–26], but to our knowledge, only a recent publication [27] on lateral heterostructures of 1H and 1T phases of MoS_2 tackles the very important aspect of the phase coexistence from the theoretical point of view. A detailed understanding of the atomic structure and properties of the interfaces between different phases of MoS_2 is of paramount importance because, besides their academic interest, heterostructures combining such phases can potentially be used in efficient nanodevices [24,25,28].

In this study we explicitly investigate the interfaces between 1H and two most relevant octahedral phases (1T and 1T') to provide an insight into the extensive reconstructions within the patches and at the boundaries and corresponding electronic structures. We have chosen to study the 1T form and the first metastable polymorph (1T'). The 1T phase is unstable from group symmetry arguments, but those analyses assume long-range periodicity. Therefore, the possibility of achieving locally metastable 1T phases coexisting as small patches in the 1H host is still open. For this purpose, we have constructed mixed phases of monolayer MoS₂ combining the ground state trigonal phase 1H acting as a matrix and one of the octahedral polymorphs 1T and 1T' as a patch. Three different supercell sizes have been considered so that the spatial region of the 1H phase between the periodic images of the patches is kept constant while the patch grows. We name the structures 1H-1PX-Y, 1P being the polymorph type (1T or 1T') that we start with, X being the size of the 1Hsupercell matrix, and Y being the size of the octahedral phase patch, in units of a hexagonal 1×1 unit cell, as Fig. 1 shows. The models have been constructed in accordance with the results reported in [27], which found that armchair boundaries lead to atomic migration and therefore have been avoided in this study, in favor of the very stable zigzag interfaces. Furthermore, and following the naming convention in the aforementioned study, each patch has Mo interfaces to the top and right and S interfaces to the bottom and left. It is important to note here that the label of 1H-1T or 1H-1T' only represents the starting geometry of the internal patch before optimization. The geometry optimization carried out for each of these supercells make extensive changes in the internal structures, such that one may not see any resemblance of the final atomic arrangements to the starting one, as discussed in detail later.

The paper is organized as follows. In the next section we discuss the effects of strain on the band structure of pure 2D phases of 1T and 1T'. Then we present a detailed analysis of the geometries and electronic structures of the mixed phases and discuss several scenarios of gap opening. Thereafter, we summarize our findings and discuss their implications. Finally, we provide the details of the computational methods. We provide Supplemental Material (SM) as an external file where we include additional figures [29].

II. METHODOLOGY

All calculations were done in the framework of density functional theory (DFT) using the projector augmented wave (PAW) [30,31] method and the Perdew-Burke-Ernzerhof (PBE) generalized-gradient approximation (GGA) [32] implemented in the Vienna *ab initio* simulation package (VASP)

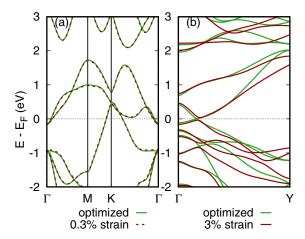


FIG. 2. Effect of strain on the band structure of the octahedral phases. Band structure of 1T (a) and 1T' (b) MoS_2 . The green solid line represents the band structure of the cell that minimized the total energy with respect to the variation of the lattice parameter in each of the phases. The dashed (solid) red line depicts the band structure of the 1T (1T') under the strain that the 1H equilibrium lattice parameter imposes.

[33–35] code, which uses a plane-wave basis. We have used a cutoff of 400 eV to truncate the plane-wave basis representing the Kohn-Sham orbitals. For the geometry optimization, the Brillouin zone was sampled by the Γ point and the atomic positions were relaxed until the force per atom is less than 0.005 eV/Å using the conjugated gradient method. A denser $3 \times 3 \times 1$ Monkhorst-Pack mesh was used for the self-consistent description of the electronic states and density of states (DOS).

III. RESULTS AND DISCUSSION

A. Strain dependence of pure phases

The three considered phases of MoS₂ have different lattice parameters. However, we have constructed the mixed phases supercells using the lattice parameter (3.190 Å) of the 1Hphase since it is the most stable and common phase and acts as the host matrix. The lattice parameters that minimize the total energy in pure 2D 1T and 1T' phases are 3.200 Å for 1T and 3.275 and 6.600 Å for 1T'. This means that the octahedral patches will be subjected to planar compressive strain, and this structural change may affect their electronic properties. The 1T phase is strained only 0.3% and it is a very stable metallic phase, so its band structure does not undergo any relevant changes, as shown in Fig. 2. However, in its optimized lattice parameter, the 1T' phase is a semimetal with bands crossing only at one reciprocal lattice point in the Brillouin zone. The nearly 3% compressive strain it is subjected to when placed in the 1H matrix is able to open up a small gap, as Fig. 2 shows, in line with what was reported previously [36].

B. Mixed phases

1. Electronic structure

The atomic positions within the mixed phase supercells, described in the Introduction, were optimized to minimize the

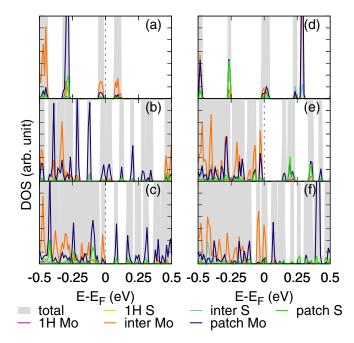


FIG. 3. Contribution to each kind of atomic type to the DOS of the six structures close to the Fermi energy. (a), (b), and (c) Represent the 1H-1T 6-3, 9-6, and 12-9 composites, respectively. (d), (e), and (f) Represent the 1H-1T' 6-3, 9-6, and 12-9 composites, respectively. The gray filled curves are the peaks from the total DOS of each structure, and the solid lines are the partial DOS of selected Mo and S atoms belonging to bulk 1H (purple and yellow, respectively), the interfacial region (orange and light blue, respectively), and the bulk region of the patch, either 1T for the left panel or 1T' for the right one (dark blue and green). Gap opening happens for (a), (c), and (e).

energy in each case. Therefore, the final structures are not composed of perfect 1H, 1T, or 1T' phases due to the reorganization of the atomic positions. First we study the density of states (DOS) of the resulting supercells, focusing on the range of energies close to the Fermi energy, shown in Fig. 3. The two kinds of starting supercells seem to behave in an opposite way after structural relaxation, while there is a clear trend towards a gap opening after relaxation: 1H-1T 6-3 and 1H-1T 12-9 show a gap opening of 84 and 85 meV, respectively, while 1H-1T 9-6 does not. On the contrary, 1H-1T' 6-3 and 1H-1T'12-9 are metallic but 1H-1T' 9-6 has a gap of 167 meV. To explain this, we compare the total energies of these structures, reported in Table I: the gapped systems have lower energy than their metallic counterparts for every size considered. This means that whichever phase the patch started in (1T or 1T'), the systems, independent of the patch size, evolve towards a

TABLE I. Total energies (in eV) and values of the gap of the mixed phases. Bold indicates the lowest energy structure of each size, which opens up a gap, included in parentheses (in meV).

	1 <i>H</i> -1 <i>T</i>	1 <i>H</i> -1 <i>T'</i>
6-3	-770.40 (84)	-770.01
9-6	-1729.82	- 1730.08 (167)
12-9	-3070.11 (85)	-3068.48

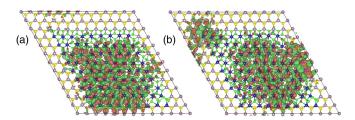


FIG. 4. Charge density of the VBM (a) and CBM (b) in 1H-1T 12-9. The red (green) color represents positive (negative) charge density with an isovalue $4.88 \times 10^{-6} e/\mathring{A}^3$.

gapped electronic structure because it is energetically favorable. This suggests that stabilizing an intrinsically metallic patch of a metastable phase of MoS₂ as an intergrowth inside a 1H matrix, independent of the patch size, is an improbable event due to energy considerations. Even when the starting atomic arrangements of the patch mimics that of the 1Tmetallic phase, we find that the lowest energy configuration invariably takes on a small gapped semiconducting nature following atomic rearrangements to accommodate the resulting strain across the interface. A plausible explanation to why the starting 1H-1T' did not result in a ground state for 6-3 and 12-9 can be related to even-odd effects. These configurations have an odd number of Mo atoms in the patch, which means that one interface Mo atom is left to dimerize with another Mo originally belonging to the 1H host. This is, for an odd size of the patch, less favorable than the starting 1H-1T arrangement in which the Mo-Mo bonds are more similar. Additionally, the emergence of a gap, even for the starting T phase patch, cannot be attributed to strain, as it has been shown in Fig. 2 that the T phase is impervious to strain and remains metallic.

We have plotted the partial densities of states of selected atoms belonging to different regions in the composite, i.e., Mo and S atoms in the 1H bulk, in the 1T or 1T' bulk region, and in the interfacial region between the two phases. As shown in Fig. 3, the states contributing mostly to the DOS close to the gap belong to Mo atoms in the bulk of the octahedral patch or Mo atoms in the interfacial region between the patch and the 1H host. The host states, both belonging to Mo and S, are further away from the composite gap, as the pure 1H phase has a gap of 1.8 eV, much larger than the small gap (tens of meVs) openings due to geometrical reconstructions. This is represented in Fig. 1 in the SM [29] for the 1H-1T and 1H-1T' 12-9 structures, in which the Mo and S states of the 1H phase are represented in purple and yellow, respectively, and are far from the gap. It is clearly seen that the electronic states emerging in the band gap of the 1H host originate from the 1T or 1T' patches. Moreover, these states are not localized on particular atoms, but they are extended over the whole patch and interfacial regions. As an example, we provide the charge density isosurfaces of the valence band maximum (VBM) and conduction band minimum (CBM) in 1H-1T 12-9 in Fig. 4.

In order to further study the effect of the reconstruction, we turn to Fig. 2 in the SM [29], which shows the total DOS of each structure before and after structural relaxation. As apparent by the changes of the peaks, the gaps open only in the relaxed structures, since all the structures were metallic before

allowing the atoms to relax to their lowest energy positions. Therefore, the gap opening can be attributed to the atomic relaxation, which in turn affects the electronic structure.

2. Structural reconstruction

Interestingly, the extent of the structural distortion is not limited to the boundary between the phases. Figures 5 and 6 illustrate this for the lowest energy structures, which show gap opening, and Fig. 3 in the SM [29] includes the high energy ones, which are metallic.

The comparison between the atomic positions before (in the gray background) and after relaxation show a clear displacement of the atoms, which is in general larger for the supercells that have a starting 1T patch, compared to 1T'. The smallest cell (6-3) patches show larger distortions than the larger ones, since the interface has a very strong effect. Reconstruction is very clear for corner atoms, specially S, as highlighted in Figs. 5(e) and 5(f) for 1H-1T, but also present in 1H-1T'. The S atoms in the 1H phase shift laterally in the 1H-1T composites [Fig. 5(g)], but this is less evident in 1H-1T' (Fig. 6), which overall suffers less changes. Nonetheless, structural distortions reach the center of the 12-9 patches, as evident in Figs. 5(h) and 6(c), in which a clear tendency towards Mo-Mo dimerization after relaxation is revealed. This is responsible for the larger degree of position changes in the 1H-1T structures compared to the 1H-1T'. The combination of short and long bonds, which is a characteristic of the 1T'phase, is also accompanied by a vertical shift in the S positions, so that the S atoms in a long Mo-Mo bond come closer to the center of the layer and the ones in a short Mo-Mo bond go further away [see Figs. 5(d) and 6(d)]. This disposition towards a 1T'-like phase in the patches of the mixed supercells can be explained by the relative stability of their pure phases, since 1T' has a much lower energy than the unstable 1T. It should be noted that, even if the general trend of the patches is to reconstruct towards a 1T'-like phase, there can be regions in which the atomic positions may locally retain a 1T type arrangement, specially in the 1H-1T interfaces like exemplified by Fig. 6(c). This could be an explanation to why the 1T phase seems to have been observed in transmission electron microscopy experiments, even if it is high in energy compared to 1T' and the ground state 1H phases.

To clearly asses how the structural reconstruction induces Mo-Mo dimerization in the mixed phases both in the boundary and center of the patches, we have plotted histograms for the nearest neighbors (NNs) of Mo (either Mo or S atoms) in Fig. 7. A histogram for the pure phases would only have counts at the characteristic NN distances that are marked with vertical lines and the spread around those distances is the mere effect of the interface distortions, which cannot be captured by theoretically studying pure phases only. In all the histograms, the first dome (2.3–2.6 Å distances) belongs to Mo-S bonds and the second one (3–3.4 Å) corresponds to Mo-Mo distances.

As expected, most of the NN distance counts in the 1H-1T supercells are concentrated around the corresponding distances for pure 1H and 1T phases. In the same way, the NN distances in 1H-1T' are centered around those of the pure 1H and 1T' phases. However, the distributions in the top row

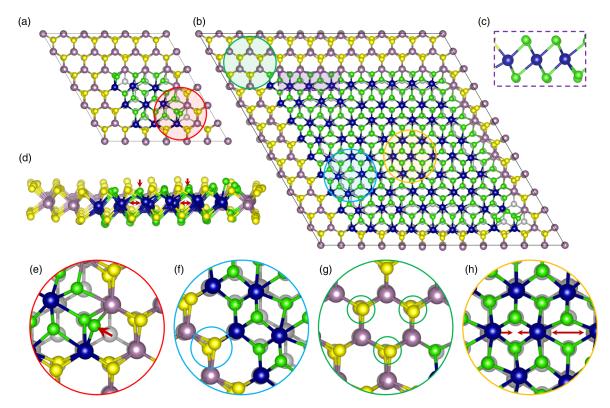


FIG. 5. Structural reconstruction in the 1H-1T low energy composites [6-3 in (a) and 12-9 in (b)]. Overlapping atomic positions before (gray) and after relaxation (following the color coding described in Fig. 1). (c) Side view of the dashed purple rectangle in (b). (d) Side view of the 1H-1T 9-6 in which the arrows point at the Mo-Mo dimerization features. (e)–(h) Zoomed regions enclosed by colored circles in (a) and (b). The arrow in (e) highlights a large atomic displacement, and the arrows in (h) show the direction of the Mo relaxation towards its dimerized positions.

(starting 1T patch) and the lower row (starting 1T' patch) are not very different for each size: 1H-1T structures show also the presence of 1T' characteristic distances, specially the 2.73 Å of the Mo-Mo short bond, in all its sizes (6-3, 9-6, and 12-9). This confirms that the reconstruction goes towards a common mixed structure with traces of Mo-Mo dimerization just as the 1T' phase has, which is what we had visually assessed before. Furthermore, it is interesting to see that even if the supercells have been designed as multiples of the 1H unit cell with the corresponding lattice parameter, the atoms in the patches are able to rearrange themselves into the 1T' bond

lengths. Therefore, given that the pure strained $1T^\prime$ phase is a small gap semiconductor, the tendency of the mixed structures to open a gap after relaxation can be rationalized. As a note, the fact that the dimerization is more apparent in the larger cells can be explained by the better statistics provided by a larger number of bonds.

IV. CONCLUSION

We have presented a detailed study of the interface reconstruction of heterostructures of different structural phases

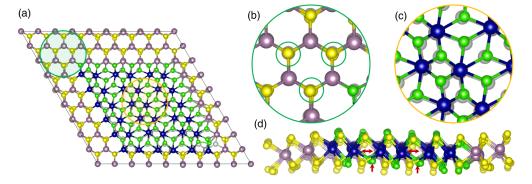


FIG. 6. Structural reconstruction in the 1H-1T' 6-3 composite. (a) Overlapping atomic positions before (gray) and after relaxation (following the color coding described in Fig. 1). (b) and (c) Zoomed regions enclosed by colored circles in (a). (d) Lateral view of (a) with arrows highlighting the features of Mo-Mo dimerization.

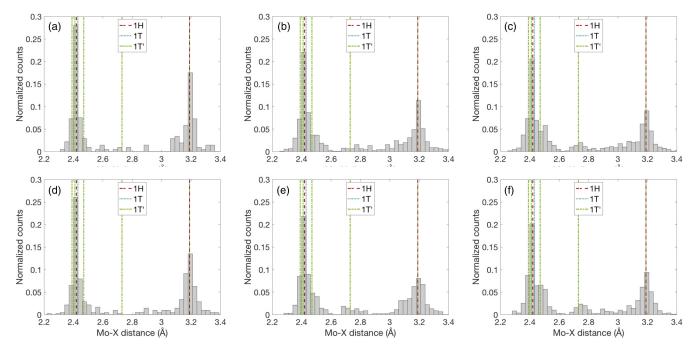


FIG. 7. Probability normalized histograms of the Mo-X distance, X being a Mo nearest neighbor atom (Mo or S). (a)–(c) 6-3, 9-6, and 12-9 1H-1T, respectively. (d)–(f) 6-3, 9-6, and 12-9 1H-1T, respectively. The vertical lines denote the pure phases nearest neighbor distances (red dashed for 1H, blue dotted for 1T, and green dash-dotted line for 1T).

of monolayer MoS_2 by density functional theory. Our study reveals that this reconstruction is essential to explain the electronic structure of the composites, as they cannot be described simply by the behavior of pure phases. The interaction between the trigonal ground state phase (1H) and any of the octahedral phases (1T) and (1T) results in a reorganization of the atoms in the metastable octahedral patch that generates a dimerization of the Mo atoms similar to what happens in the (1T) structure. This structural change is responsible for the opening of a small gap in the density of states of the composite structure, leading to a semiconducting nature. The importance of the structural reconstruction and gap opening in the samples having coexisting phases of (1T)000 experimental reports on the subject.

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^[1] A. K. S. Novoselov, A. K. Geim, S. V. Morozov, D. A. Jiang, Y. Zhang, S. V. Dubonos, I. V. Grigorieva, and A. A. Firsov, Science 306, 666 (2004).

^[2] B. Pal, A. Singh, G. Sharada, P. Mahale, A. Kumar, S. Thirupathaiah, H. Sezen, M. Amati, L. Gregoratti, U. V. Waghmare, and D. D. Sarma, Phys. Rev. B 96, 195426 (2017).

^[3] A. Splendiani, L. Sun, Y. Zhang, T. Li, J. Kim, C. Y. Chim, G. Galli, and F. Wang, Nano Lett. 10, 1271 (2010).

^[4] K. F. Mak, C. Lee, J. Hone, J. Shan, and T. F. Heinz, Phys. Rev. Lett. 105, 136805 (2010).

^[5] Y. Sun, S. C. Wu, M. N. Ali, C. Felser, and B. Yan, Phys. Rev. B 92, 161107(R) (2015).

^[6] E. Devers, P. Afanasiev, B. Jouguet, and M. Vrinat, Catal. Lett. 82, 13 (2002).

^[7] W. Zhong, W. Tu, S. Feng, and A. Xu, J. Alloys Compd. 772, 669 (2019).

^[8] G. Salomon, A. W. De Gee, and J. H. Zaat, Wear 7, 87 (1964).

^[9] M. Chhowalla and G. A. Amaratunga, Nature (London) 407, 164 (2000).

^[10] R. G. Dickinson and L. Pauling, J. Am. Chem. Soc. 45, 1466 (1923).

^[11] I. Song, C. Park, and H. C. Choi, RSC Adv. 5, 7495 (2015).

^[12] G. Eda, H. Yamaguchi, D. Voiry, T. Fujita, M. Chen, and M. Chhowalla, Nano Lett. 11, 5111 (2011).

^[13] D. Pariari and D. D. Sarma, APL Mater. 8, 040909 (2020).

^[14] T. Böker, R. Severin, A. Müller, C. Janowitz, R. Manzke, D. Voß, P. Krüger, A. Mazur, and J. Pollmann, Phys. Rev. B 64, 235305 (2001).

- [15] B. Radisavljevic, A. Radenovic, J. Brivio, V. Giacometti, and A. Kis, Nat. Nanotechnol. 6, 147 (2011).
- [16] Q. H. Wang, K. Kalantar-Zadeh, A. Kis, J. N. Coleman, and M. S. Strano, Nat. Nanotechnol. 7, 699 (2012).
- [17] D. Pariari, R. M. Varma, M. N. Nair, P. Zeller, M. Amati, L. Gregoratti, K. K. Nanda, and D. D. Sarma, Appl. Mater. Today 19, 100544 (2020).
- [18] F. Wypych and R. Schöllhorn, J. Chem. Soc. Chem. Commun. 19, 1386 (1992).
- [19] M. Acerce, D. Voiry, and M. Chhowalla, Nat. Nanotechnol. 10, 313 (2015).
- [20] M. Calandra, Phys. Rev. B 88, 245428 (2013).
- [21] G. Eda, T. Fujita, H. Yamaguchi, D. Voiry, M. Chen, and M. Chhowalla, ACS Nano 6, 7311 (2012).
- [22] Y. C. Lin, D. O. Dumcenco, Y. S. Huang, and K. Suenaga, Nat. Nanotechnol. 9, 391 (2014).
- [23] M. M. Ugeda, A. Pulkin, S. Tang, H. Ryu, Q. Wu, Y. Zhang, D. Wong, Z. Pedramrazi, A. Martín-Recio, Y. Chen, F. Wang, Z. X. Shen, S. K. Mo, O. V. Yazyev, and M. F. Crommie, Nat. Commun. 9, 3401 (2018).

- [24] K.-A. N. Duerloo, Y. Li, and E. J. Reed, Nat. Commun. 5, 4214 (2014).
- [25] W. Zhao and F. Ding, Nanoscale 9, 2301 (2017).
- [26] Q. Jin, N. Liu, B. Chen, and D. Mei, J. Phys. Chem. C 122, 28215 (2018).
- [27] X. Zou, Z. Zhang, X. Chen, and B. I. Yakobson, J. Phys. Chem. Lett. 11, 1644 (2020).
- [28] R. Kappera, D. Voiry, S. E. Yalcin, B. Branch, G. Gupta, A. D. Mohite, and M. Chhowalla, Nat. Mater. 13, 1128 (2014).
- [29] See Supplemental Material at http://link.aps.org/supplemental/ 10.1103/PhysRevB.102.165412 for details on DOS and atomic positions before and after relaxation.
- [30] P. E. Blöchl, Phys. Rev. B 50, 17953 (1994).
- [31] G. Kresse and D. Joubert, Phys. Rev. B 59, 1758 (1999).
- [32] J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. 77, 3865 (1996).
- [33] G. Kresse and J. Hafner, Phys. Rev. B 47, 558 (1993).
- [34] G. Kresse and J. Furthmüller, Phys. Rev. B 54, 11169 (1996).
- [35] G. Kresse and J. Furthmüller, Comput. Mater. Sci. 6, 15 (1996).
- [36] T. Hu, R. Li, and J. Dong, J. Chem. Phys. 139, 174702 (2013).